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This Ring Tester is an inexpensive and effective way to test any high-Q inductive component. It is especially useful for doing a

ring tester assembly manual

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Transformers, B+ coils and other high frequency inductive

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components like
Horizontal deflection
yoke windings and
SMPS transformers.
In other words the
Blue Ring Tester is
also a Coil Tester or a
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blueringtesterresult

Page 6 Strip and tin
the one end of the
test lead ends. Put
the two grommets into

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the test lead holes in the enclosure end panel. Then feed the test leads through the grommets. The outside of the panel is the finished side.

Page 7 Test lead installation Strip approximately 1/8" (2 mm) from one end of the test leads and tin using rosin core ...

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*ANATEK BLUE ESR
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In this video I give an overview of the Blue Ring Tester by Anatek Corporation. The assembly is shown in its entirety, followed by a short review. Quick Links...

*Anatek Blue Ring
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color to alert you.
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connector pins for the
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Capacitor Lab and
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Meter – Alltronics LLC*

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inductive component.
Features of the Blue
Ring Tester : Supplied
in kit form with full

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Ring Tester
Manual
assembly with
calibration instructions
and manual online
One foot test leads
made with super
flexible test lead wire
with crocodile clips

*AnaTek Blue Ring
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In 2007 Anatek
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Parker in Australia to produce the distinctive and popular "Blue" ESR and Ring Tester Meters. In 2014 Anatek was acquired by Alltronics LLC and we continue to proudly offer the "Blue" range of component testers and also sell many other new and surplus

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to Run a Test After setup, push the front button on your Ring Doorbell indoors near your router to see video from your Ring Doorbell in the Ring app. Then test your Ring Doorbell outside. If video quality from your Ring Doorbell is not as good outdoors as it was indoors, you may need to move

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An Excellent Book in
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Accordance with the latest syllabus for Class-11 Prescribed by CBSE/NCERT and Adopted by Various State Education Boards. (A) Basic Laboratory

Techniques – 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass

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Ring, 4. To bore a
cork and fit a glass
tube into it. (B)

Characterisation and
Purification of
Chemical

Substances- 1. To
determine the melting
point of the given
unknown organic
compound and its
identification (simple
laboratory technique),
2. To determine the

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boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum [$K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$] from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample,

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5. To prepare pure crystals of benzoic acid from a given impure sample. (C)
Measurement of pH Values
1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH

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values of solutions of
strong acid (HCl) and
weak acid

(CH₃COOH) of same
concentration, 3. To
study the pH change
in the titration of
strong base Vs.

strong acid by using
universal indicator
paper, 4. To study the
pH change by
common ion

(CH₃COO⁻ ion) in

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case of weak acid
(CH_3COOH), 5. To
determine the change
in pH value of weak
base (NH_4OH) in
presence of a
common ion (NH_4^+),
(D) Chemical
Equilibrium 1. To
study the shift in
equilibrium between
ferric ions and
thiocyanate ions by
changing the

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concentrations of either of the ions, 2. To study the shift in equilibrium between $[\text{Co}(\text{H}_2\text{O})_6]^{2+}$ and Cl^- ions by changing the concentrations of either of the ions, (E) Quantitative Analysis

1. To prepare M/10 oxalic acid solution by direct weighing method, 2. To prepare M/10 solution of

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sodium carbonate by direct weighing method, 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a

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standard N/10 or

M/20 sodium

carbonate solution,

(F) Qualitative

Analysis 1. Analysis

of Anions, 2. Analysis

of Cations (G)

Detection of Elements

in Organic

Compounds 1.To

detect the presence of

nitrogen, sulphur and

halogens in a given

organic compound by

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Lassaigne's test, 2.

To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by

Lassaigne's test
INVESTIGATORY
PROJECTS (A)

Checking of Bacterial Contamination in Water 1. To check the bacterial

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contamination in drinking water by testing sulphide ions
(B) Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3.To purify water by distillation method, 4.To purify water by reverse

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osmosis technique.

5.To purify water by GAC method, 6.To purify water by bleach treatment, 7.To purify water by oxidising agent, 8.To purify water by ozone treatment method. (C)

Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various

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Soaps 1. To compare the foaming capacity of different washing soaps, 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable

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1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1. To compare the tensile strength of

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natural fibres and
synthetic fibres, 2.To
study the effect of
acids and bases on
tensile strength of
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