

## Chapter 13 Chemistry Answers

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Reaction = 2 x reverse of original reaction. So  $K_2 = (1/K_1)^2 = (1/0.447)^2 = 5.00$ . 3.  $K = \frac{[NH_3]^2}{[N_2] \times [H_2]^3}$ ; more product made. Increasing the pressure increases the concentration; counteract this by shifting to the side with fewer moles.

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## *13.1 Chemical Equilibria - Chemistry*

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Answer/Explanation. Answer: d. Explanation: 4. An organic compound C<sub>7</sub>H<sub>9</sub>N forms clear solution when dissolved in KOH after reacting with C<sub>6</sub>H<sub>5</sub>SO<sub>2</sub>Cl, 'A' on diazotisation at 0°C and then reaction with ?-naphthol gives orangish red dye. 'A' on electrophilic substitution gives single product. 'A' is.

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## *Holt Chemistry Chapter 13 Answer Key*

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