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Answers 14. Multiple Choice

Questions 14-17

The spontaneous reaction that occurs when the cell in the picture operates is as follows: $2\text{Ag}^+ + \text{Cd}(s) \rightarrow 2\text{Ag}(s) + \text{Cd}^{2+}$ (A) Voltage increases. (B) Voltage decreases but remains > zero.

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Questions (Type-II)

Note : In the

following questions

two or more than

two options may be

correct. The

positive value of

the standard

electrode potential

of $\text{Cu}^{2+} / \text{Cu}$

indicates that

_____. (i) this redox

couple is a stronger

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Multiple Choice

Questions

Answers

reducing agent than the H^+ / H_2 couple. (ii) this redox couple is a stronger oxidising agent than H^+ / H_2 .

Class 12 Important Questions for

Chemistry –

Electrochemistry

31. During

electrolysis of

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Multiple Choice

Questions

Answers

water the volume of oxygen liberated is 2.24 dm^3 . The volume of hydrogen liberated, under same conditions will be :
 2.24 dm^3 . 1.12 dm^3 .
 4.48 dm^3 . 0.56 dm^3 . Answer. 32.

In Down's method for the extraction of sodium, the melting point of the

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electrolyte is
lowered by adding.

Questions

Answers

Important
Questions of
Electrochemistry
for NEET Chemistry

...

Answers: 1. (d) 2.
(e) 3. (a) 4. (e) 5.
(d) 6. (a) 7. (d) 8.
(b) 9. (d) 10. (c) 11.
(b) 12. (c) 13. (b)
14. (c) 15. (c) 16.

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(d) 17. (e) 18. (b)
19. (e) 20. (c) 21.
(c) 22. (d)

Answers

Sample Questions -
Chapter 21

If $E^\circ \text{Fe}^{2+}/\text{Fe} = -0.441 \text{ V}$ and E°

$\text{Fe}^{2+}/\text{Fe}^{2+} = 0.771 \text{ V}$, the

standard EMF of
the reaction, $\text{Fe} + 2\text{Fe}^{3+} \rightarrow 3\text{Fe}^{2+}$
will be. (a) 1.212 V.

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(b) 0.111 V. (C)

0.330 V. (d) 1.653

V. Answer. Answer:

a. We hope the

given Chemistry

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Multiple choice questions Try the following multiple choice questions to test your

knowledge of this chapter. For each question there is one correct

answer. The periodic table, physical constants and relative atomic

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these problems are
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Chapter 21

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Electrochemistry

Examples of
Multiple Choice

Questions: 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____ ; the chemical change that occurs at this electrode is called _____. (a) anode, oxidation

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(b) anode, Choice

reduction (c)

cathode, oxidation

(d) cathode,

reduction

Electrochemistry

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Questions - Exam

Answers Free

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that best
completes the
statement or

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Answers the Multiple Choice Questions Answers

question. _____ 1.

The half-reaction that occurs at the cathode during the electrolysis of molten sodium bromide is

- _____ . a. $+ 2e^- + 2Br^- \rightarrow Br_2$. b. $+ 2e^- + Br_2 \rightarrow 2Br^-$. c. $+ e^- + Na^+ \rightarrow Na$. d.

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Answer. Answer:

(b) 2. An increase

in the conductivity

equivalent of a

solid electrolyte

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with dilution is primarily due to.

(a) increased ionic mobility of ions. (b)

100 percent

electrolyte

ionisation with

natural dilution. (c)

increase in both ion

numbers and ionic

mobility. (d) A rise

in ion counts.

MCQ on

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1. The aqueous solution of which of the following compounds is the best conductor of electric current?

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Answers

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