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None of these. Answer: A Learning Objective: 2.2 Demonstrate how to read and draw bond-line structures through converting other styles of molecular representation into bond-line structures and vice versa Difficulty: Easy. 35. Which of the following is the correct bond-line structure for (CH.

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Organic Chemistry Questions The Covalent Bond 1. The hybridization of the central carbon in CH₃C≡N and the bond angle CCN are a. sp², 180°. b. sp, 180°. c. sp², 120°. d. sp³, 109°. 2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

~~General Organic Chemistry Questions~~

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